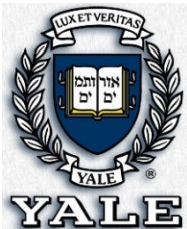


The Promise of Forward Osmosis

Menachem Elimelech



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Architectural Engineering
Korea University



Waterworks Research Institute Seoul Metropolitan Government

August 13, 2009

Motivation: Augment Water Supply

- Droughts and water shortages
- Need to increase water supply by producing new water
- Viable options for new water:
 - Wastewater reuse
 - Desalination of sea/brackish water

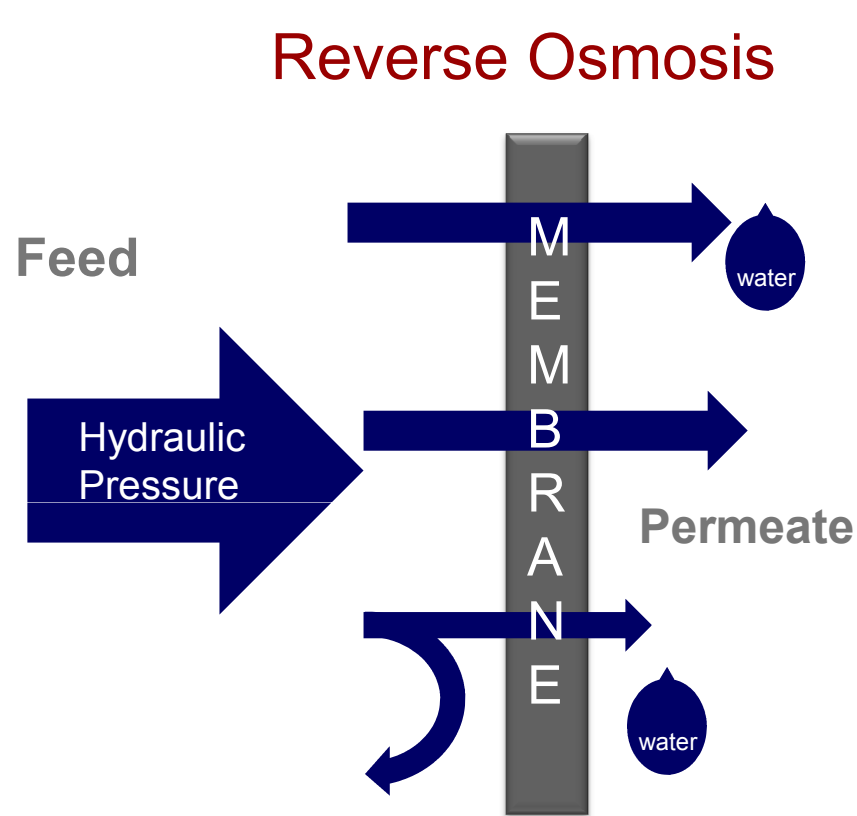


Need for Sustainable Technologies

- Develop water/wastewater treatment technologies that
 - Use less energy
 - Require less chemicals
 - Have lower impact on the environment
- Osmotically-driven membrane processes, or **forward osmosis (FO)**, may be a promising option

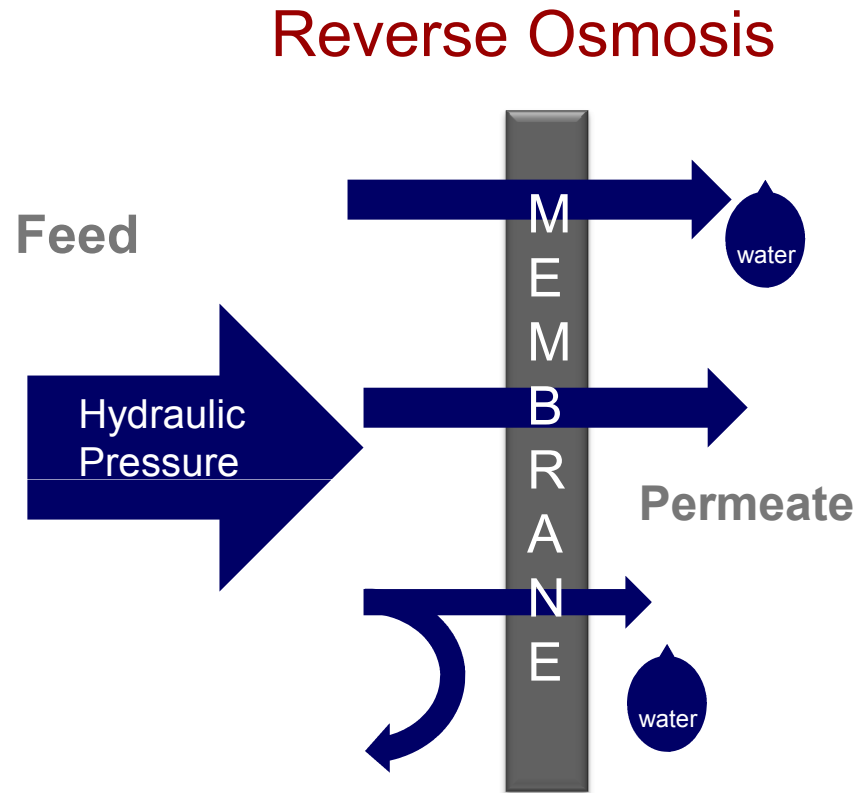


What is Forward Osmosis?

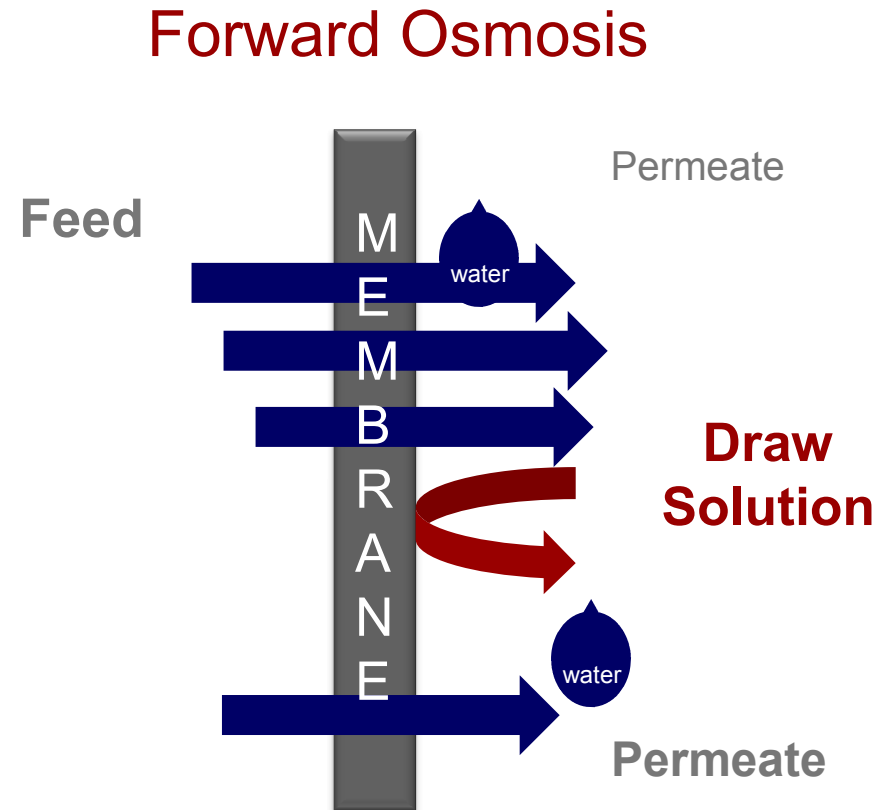


$$J_w = A(\Delta P - \Delta \Pi_m)$$

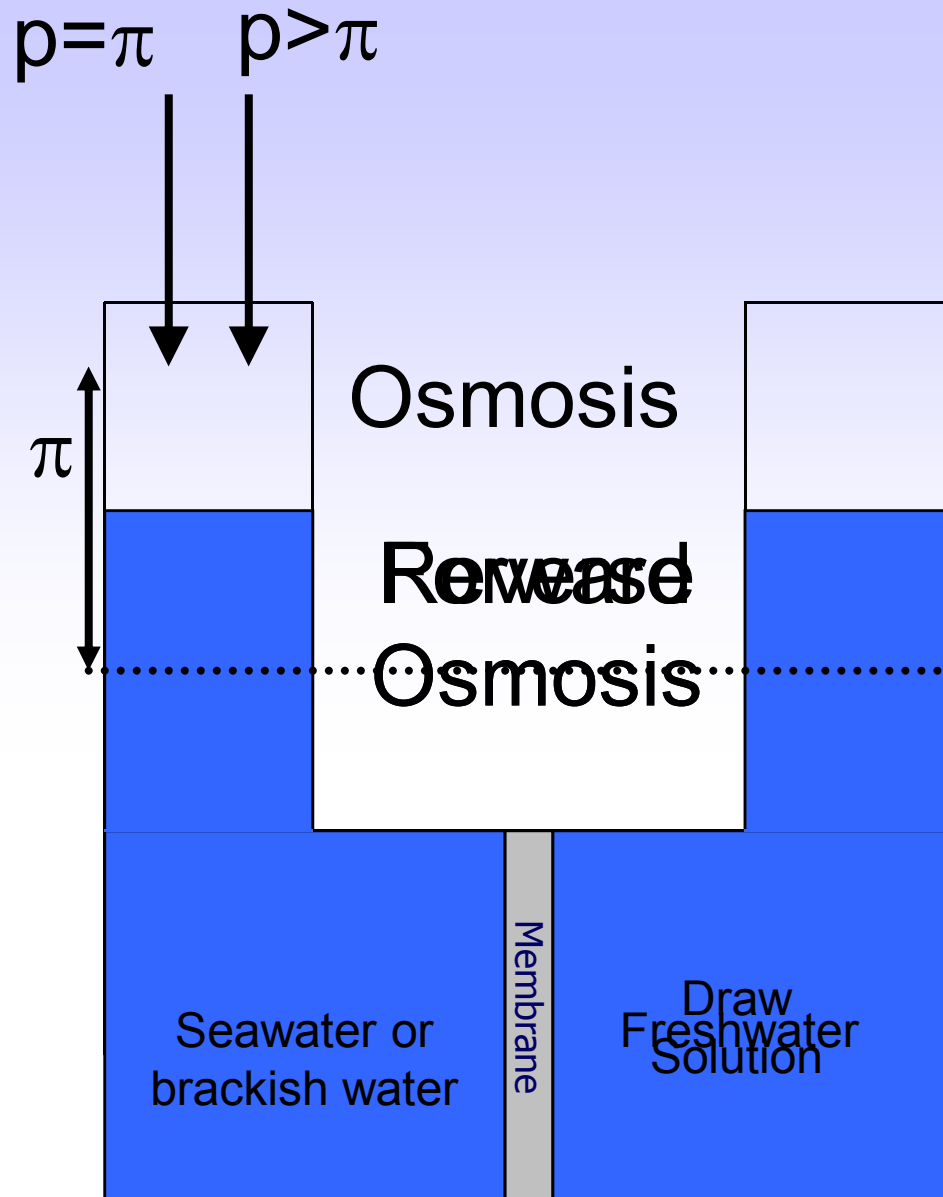
What is Forward Osmosis?



$$J_w = A(\Delta P - \Delta \Pi_m)$$

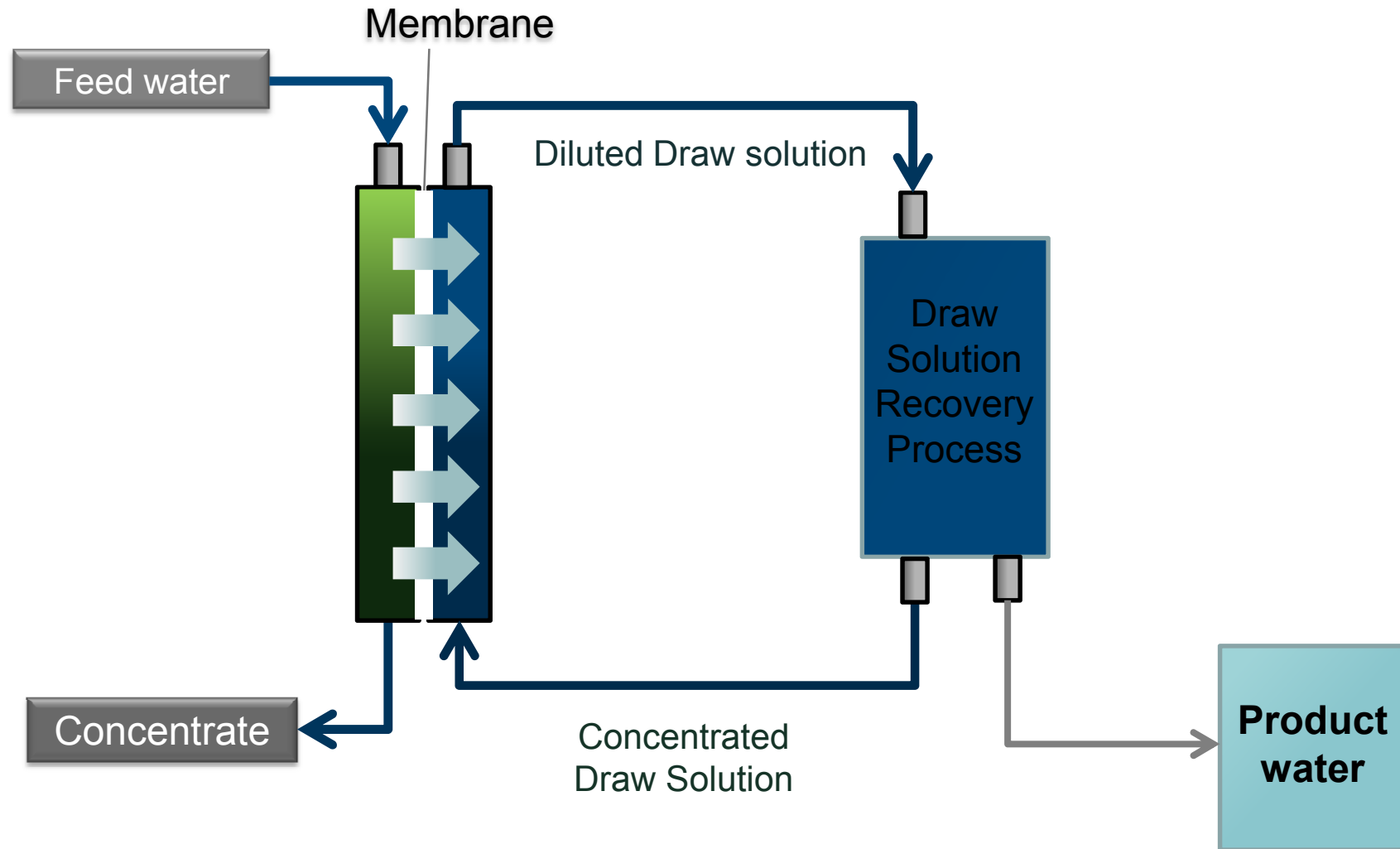


$$J_w = A\Delta \Pi_m$$



Forward Osmosis

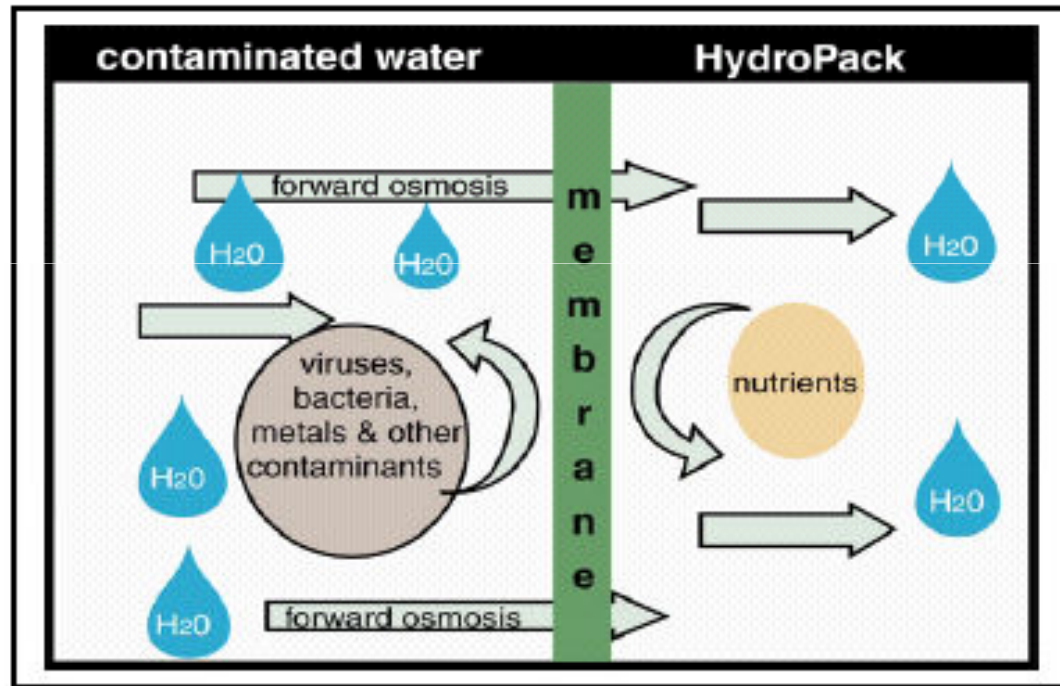
Forward Osmosis Process



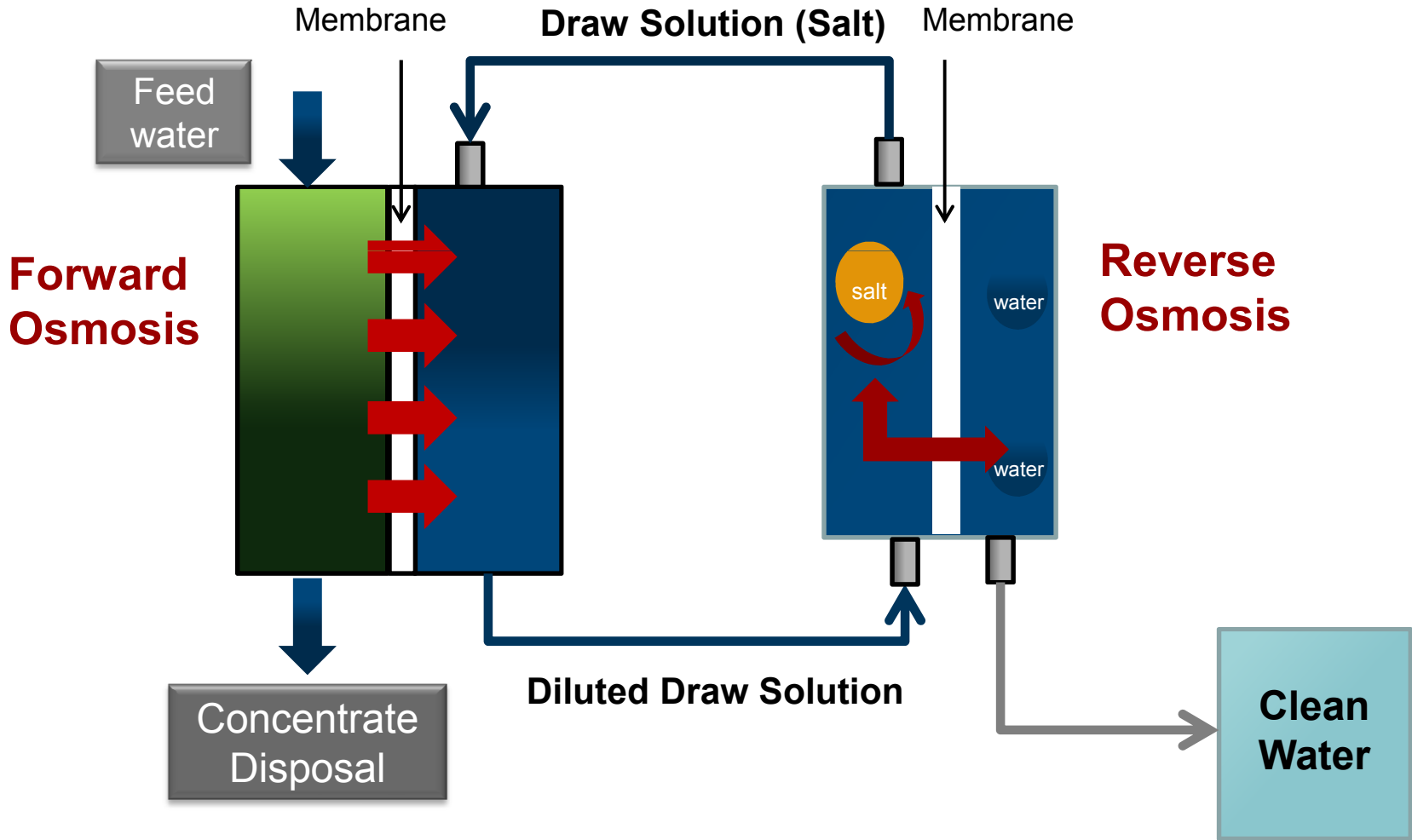


Applications of Forward Osmosis

Hydration Bags



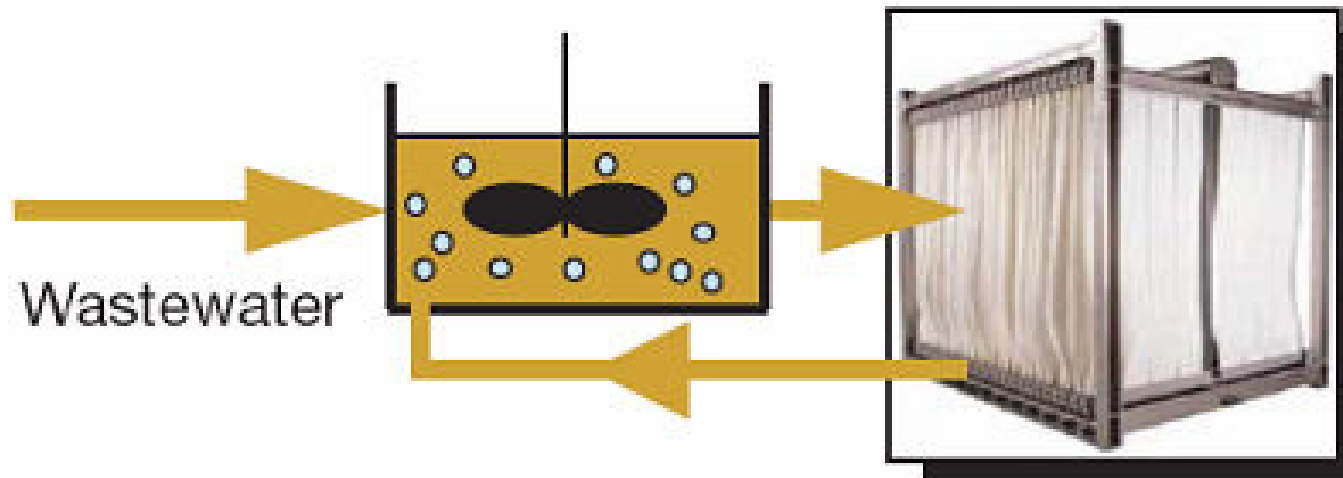
Wastewater Treatment with Forward Osmosis



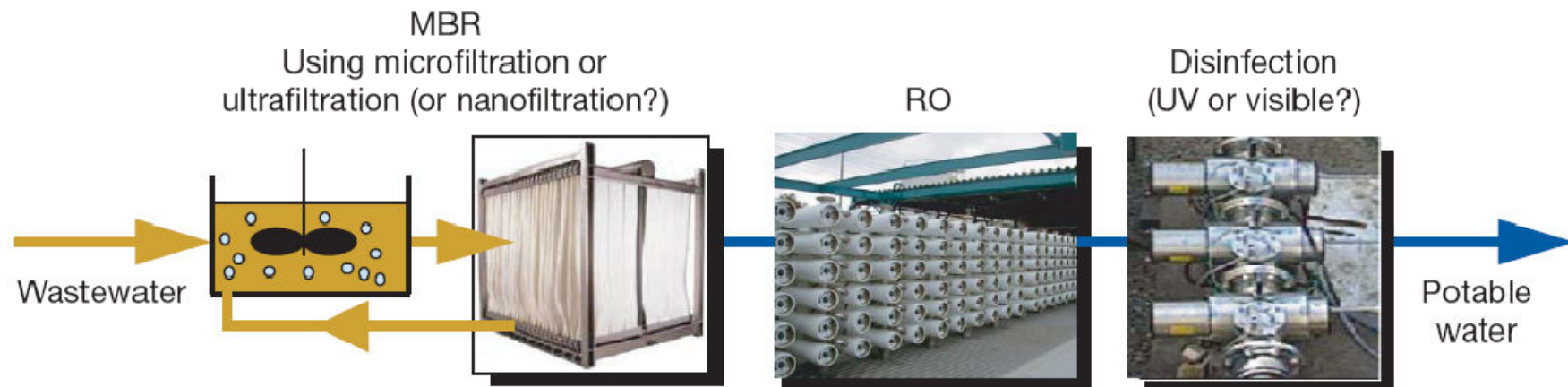
Membrane Bioreactor (MBR) for Wastewater Treatment

MBR

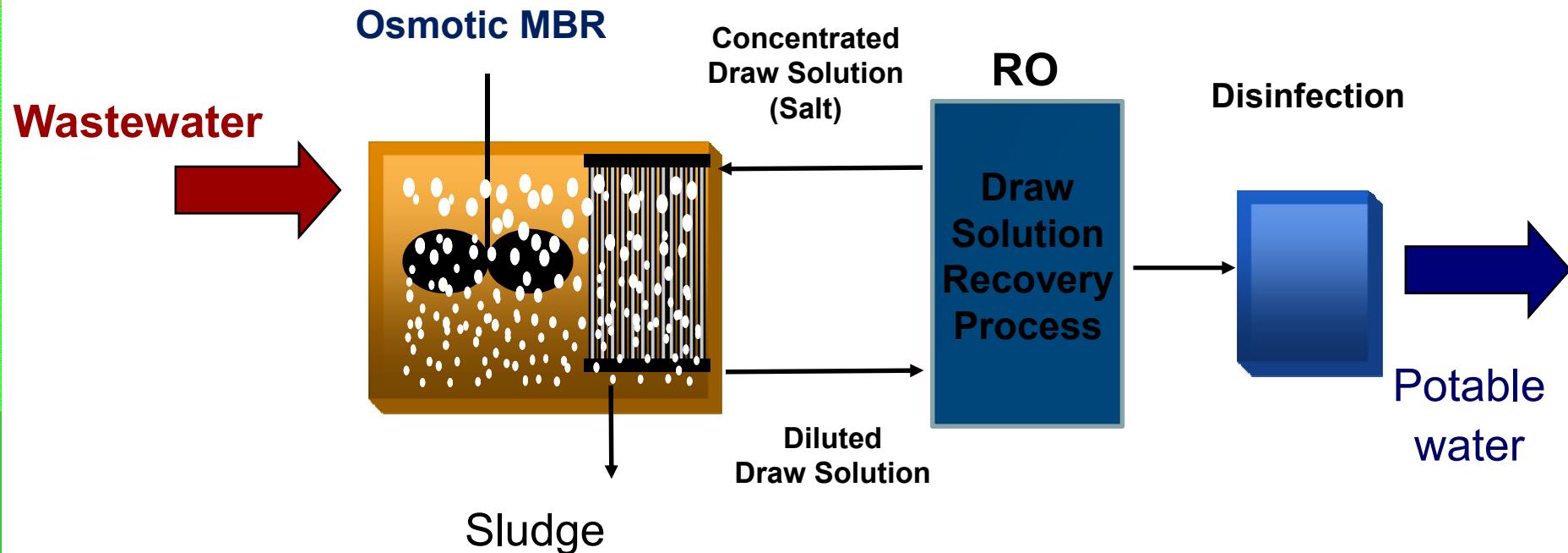
Using microfiltration or ultrafiltration (or nanofiltration?)



Wastewater Reuse: Membrane Bioreactor (MBR)-RO System



Osmotic MBR-RO: Multiple Barrier Wastewater Treatment



M1 This you can modifu as you wish. See original paper...

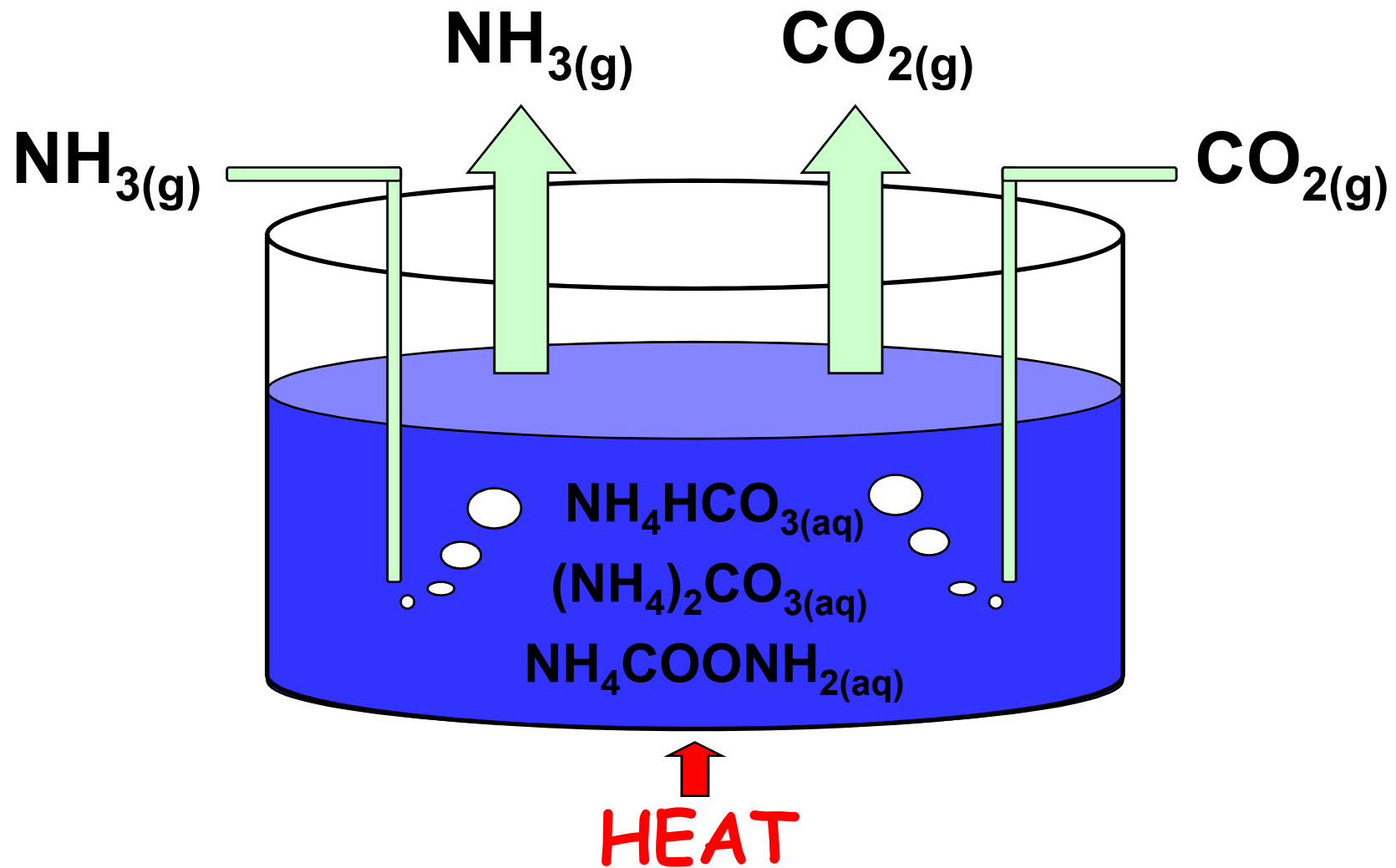
May use similar style as previous slide
Meny Elimelech, 2009-07-31



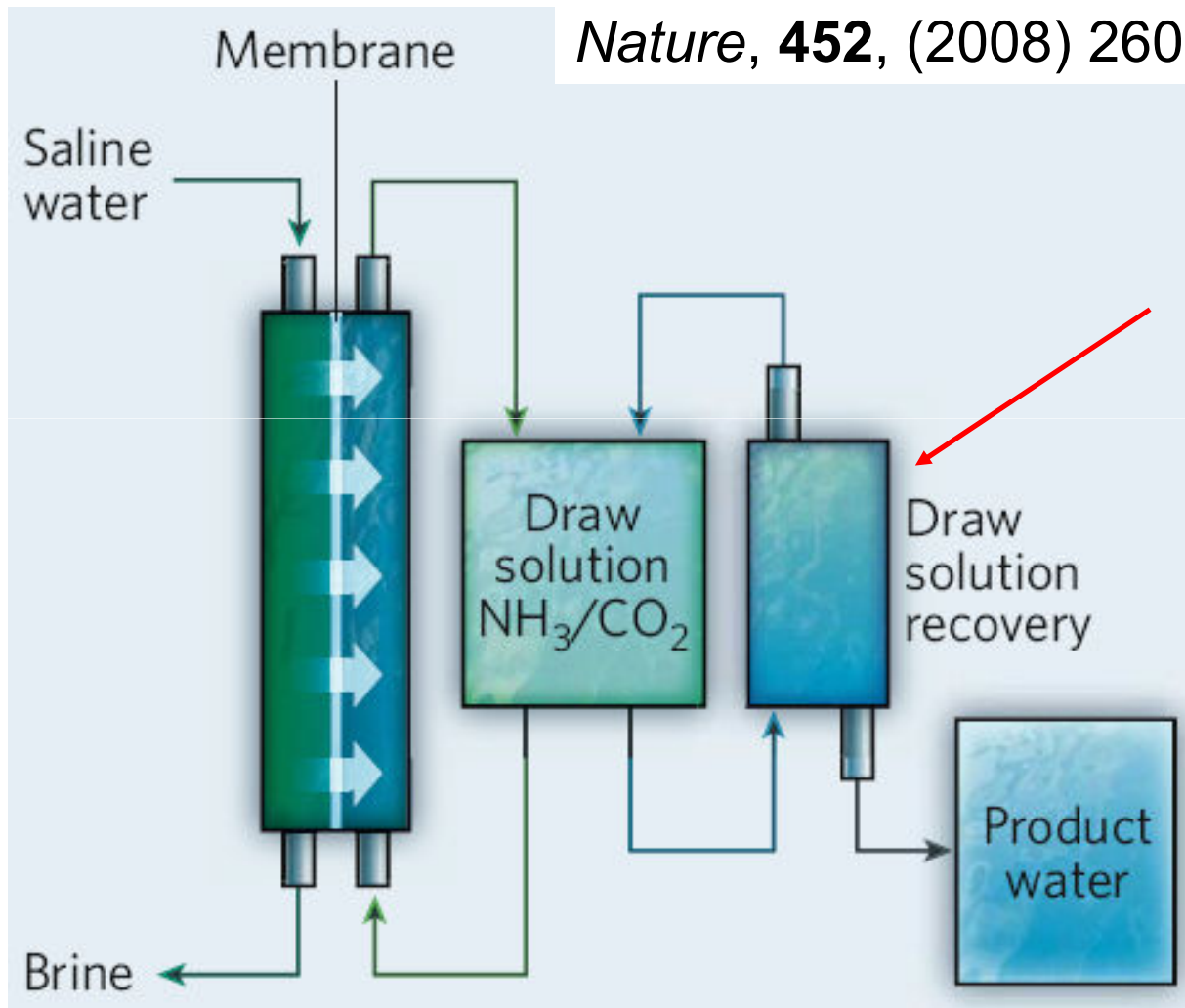
Desalination by Forward Osmosis: The Ideal Draw Solution

- Highly soluble solution to generate high osmotic pressure gradient
- Recoverable and recyclable
- Soluble species should not pass through the membrane

NH_3/CO_2 Draw Solution



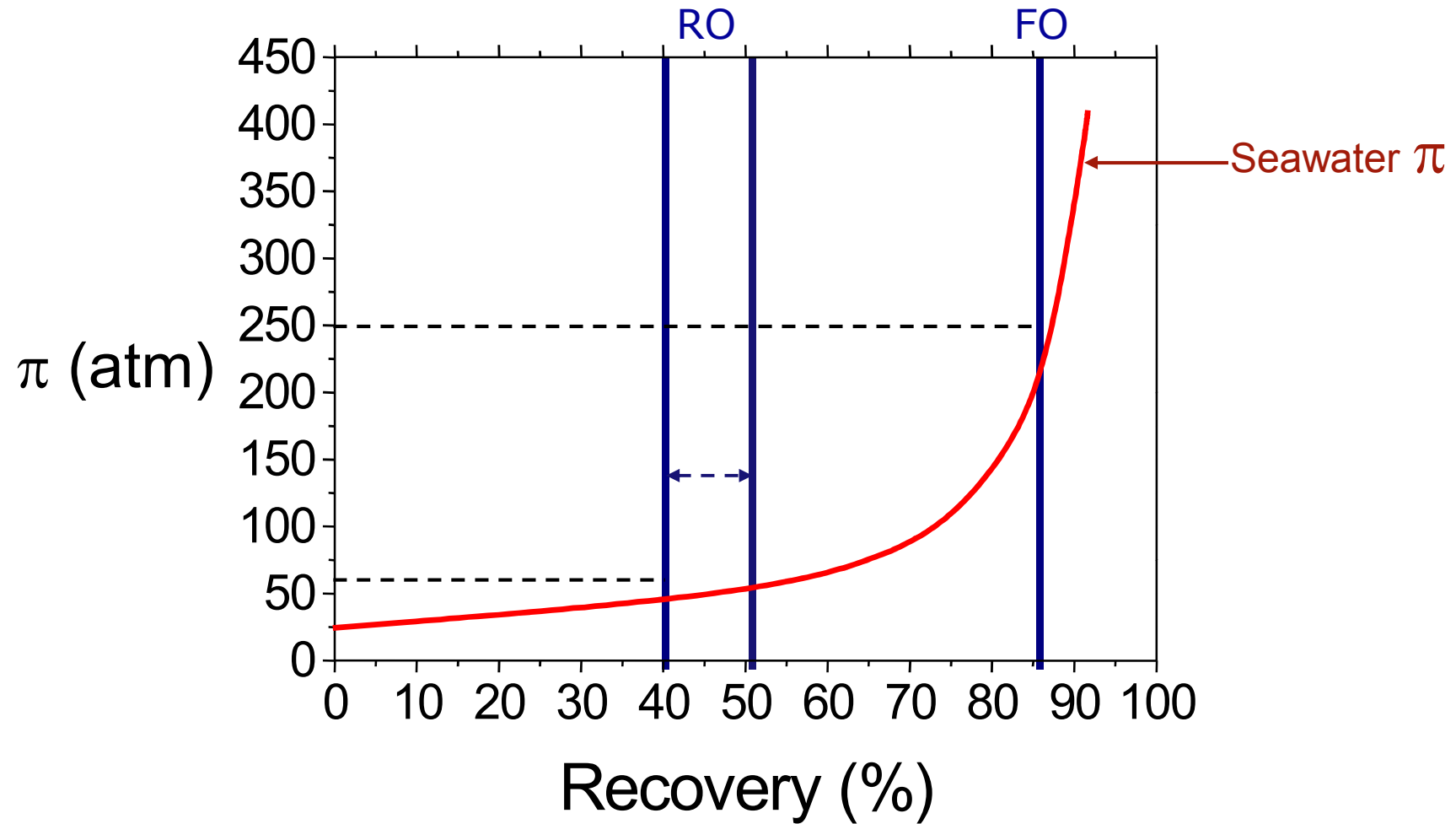
The Ammonia-Carbon Dioxide FO Desalination Process



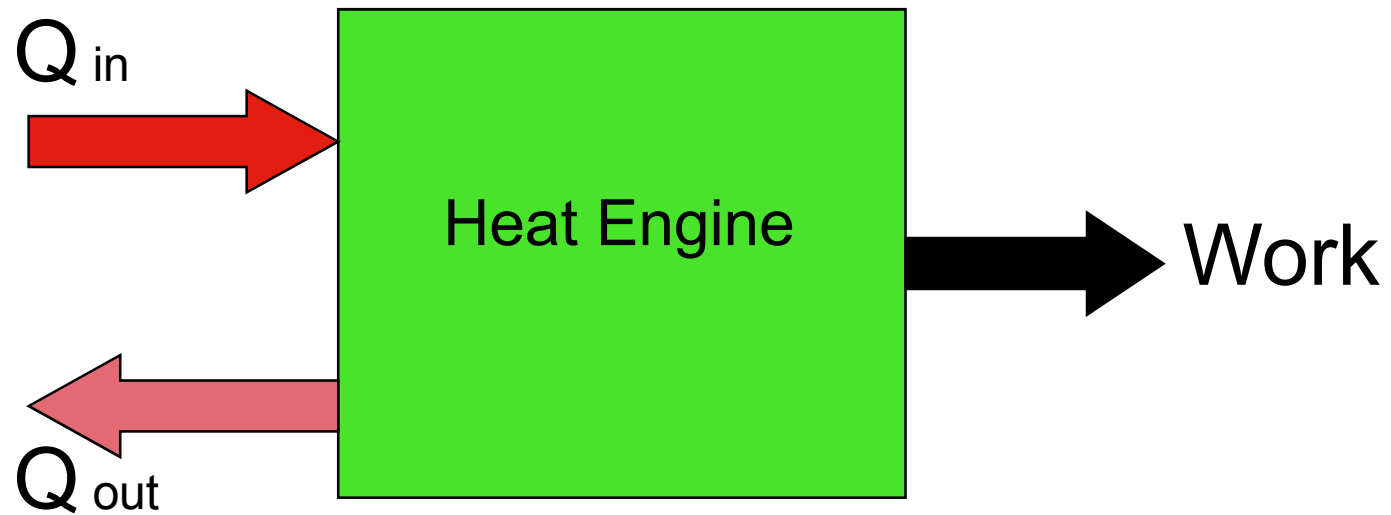
Nature, **452**, (2008) 260

Desalination, **174** (2005) 1-11.

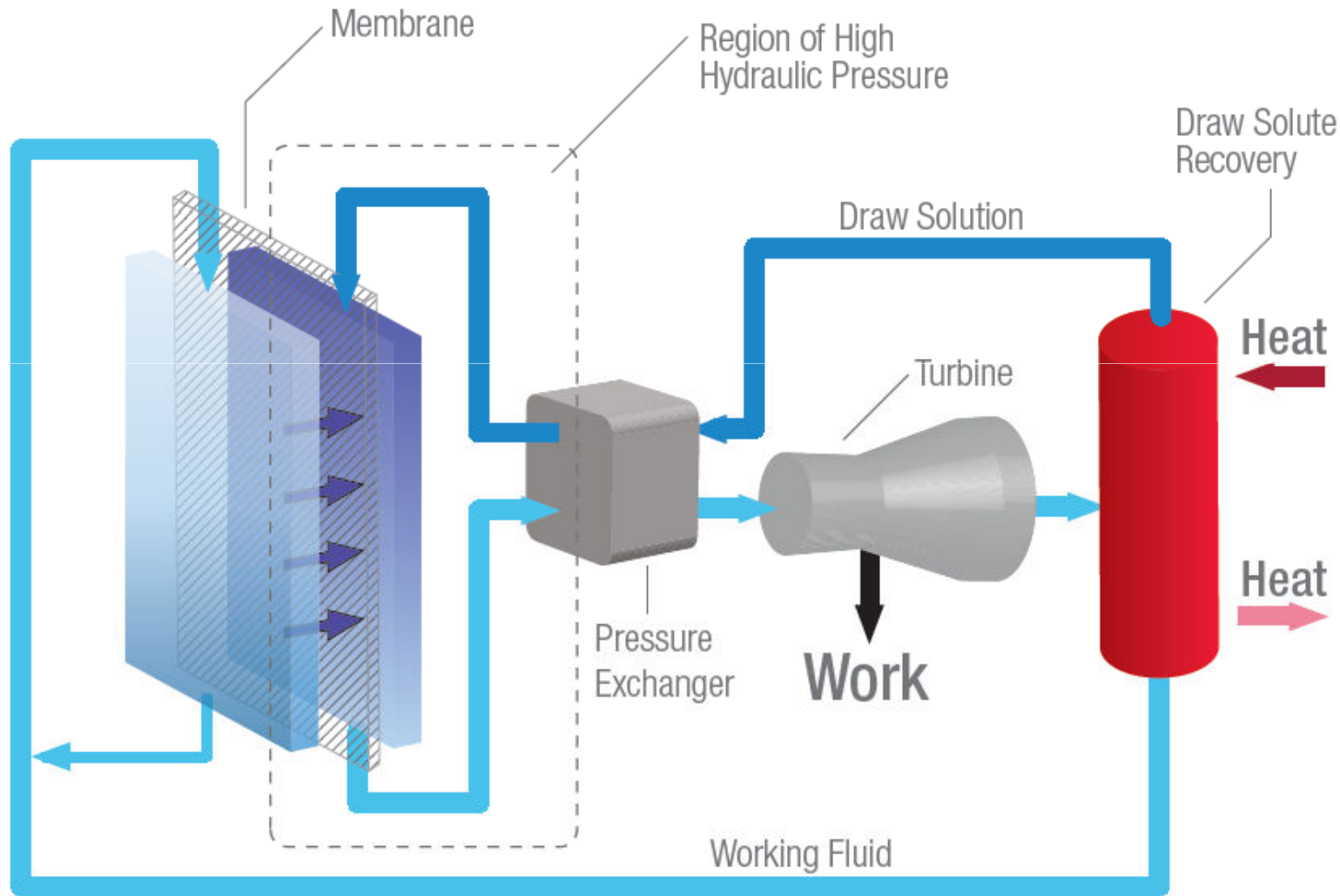
High Water Recovery with FO



Osmotic Heat Engine



NH₃/CO₂ Osmotic Heat Engine: Closed Loop PRO

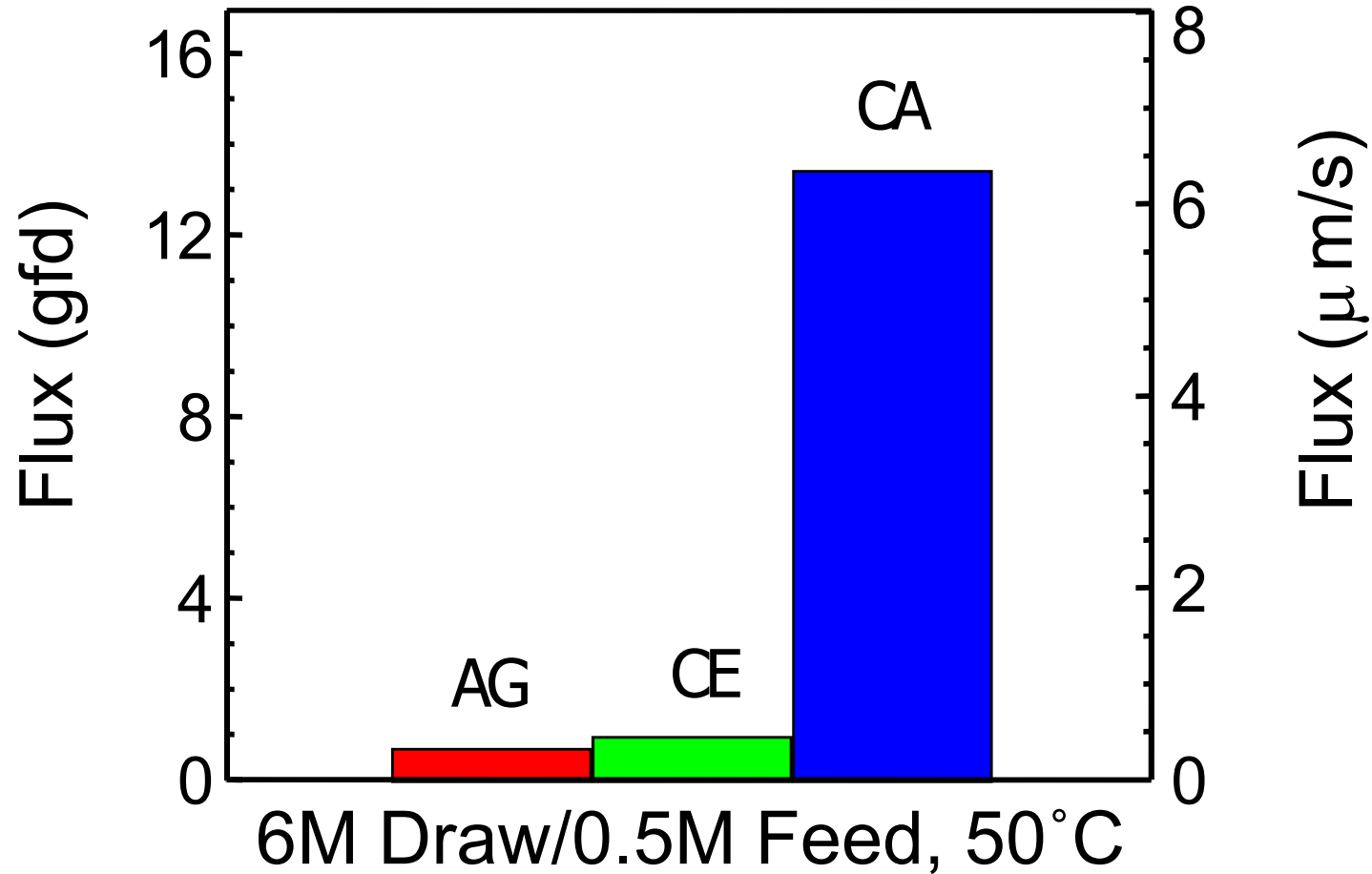


JMS, **305** (2007) 13-19; *ES&T*, **42** (2008) 8625-8629.



Water Flux in Forward Osmosis

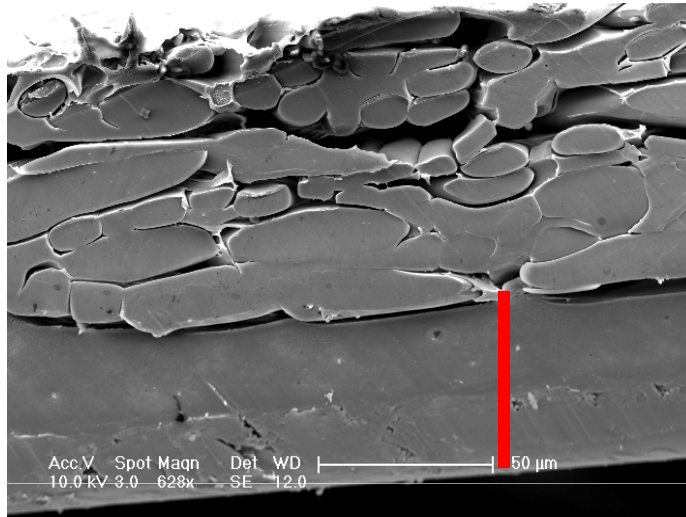
Effect of Membrane Design on FO Water Flux



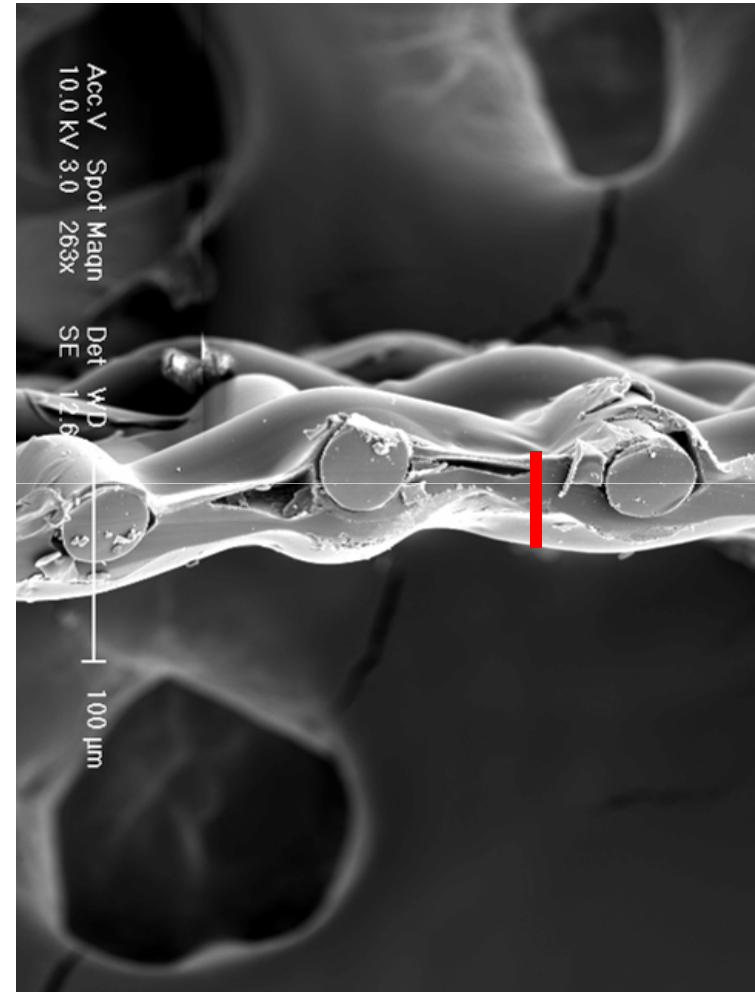
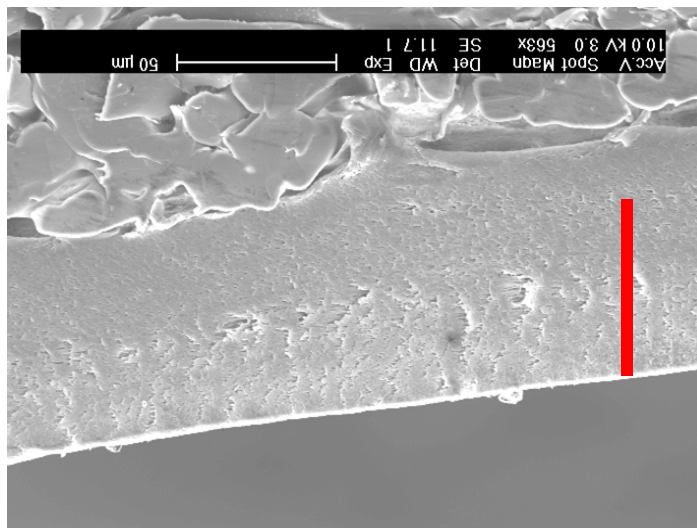
Desalination, 174 (2005) 1-11.

SEM Cross Sections

CE
(RO)



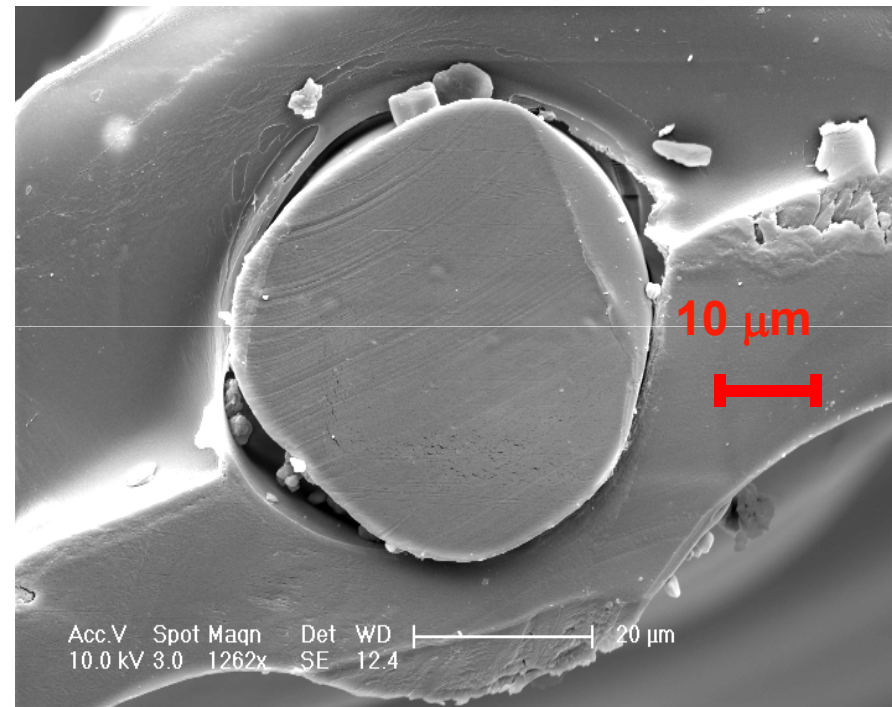
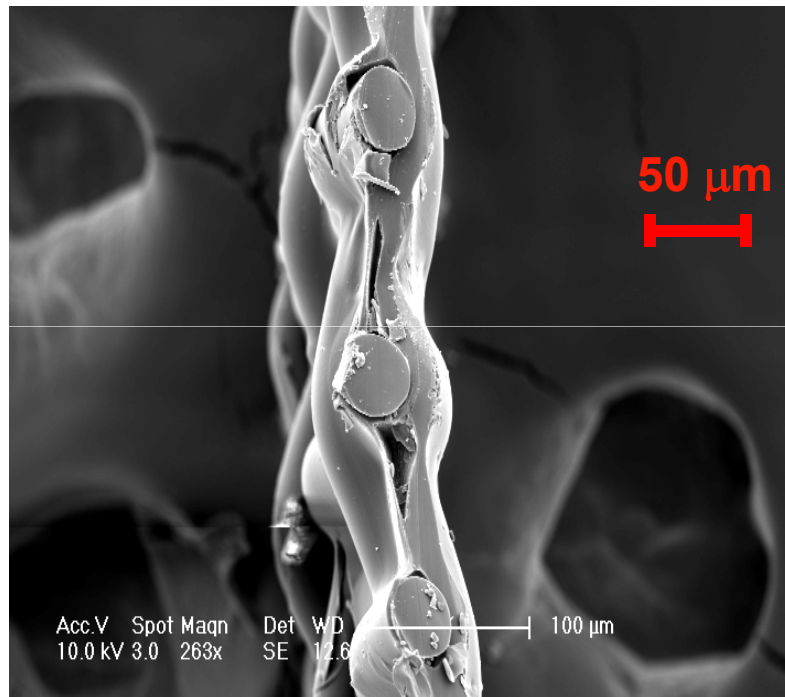
AG
(RO)



CA (FO)

Red lines indicate 50 μm

Forward Osmosis Membrane

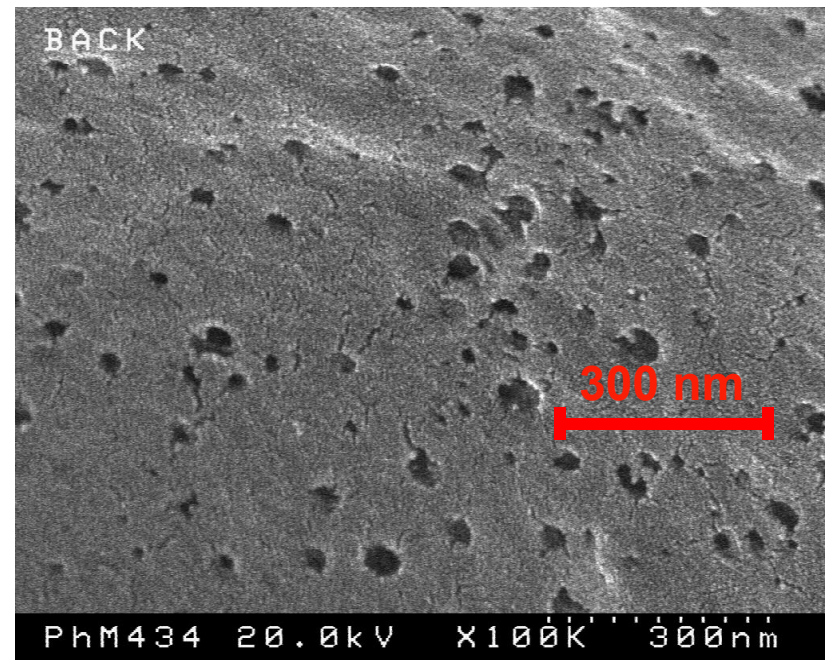


Forward Osmosis Membrane

Active Layer

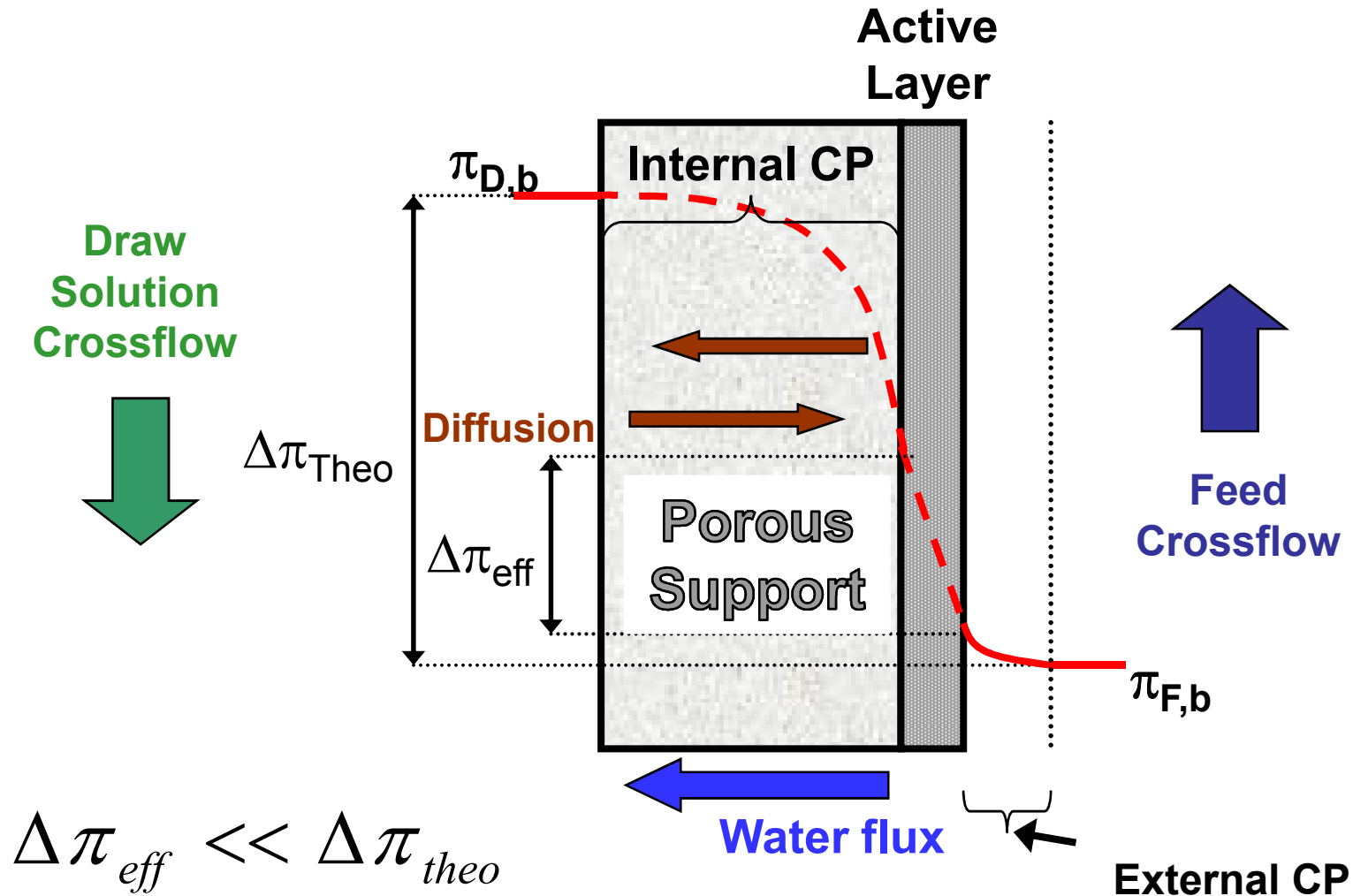


Backing Layer



- Asymmetry observed at high magnification

Major Challenge: Internal Concentration Polarization



Modeling Internal CP

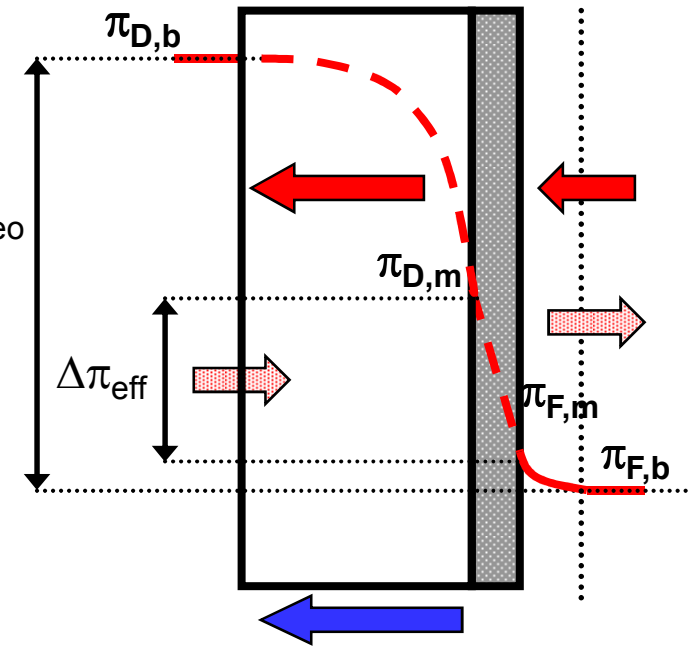
Draw Side

Feed Side

$$\frac{\pi_{D,m}}{\pi_{D,b}} = \exp(-J_W K)$$

$$\frac{\pi_{F,m}}{\pi_{F,b}} = \exp(J_W / k_F) \Delta\pi_{\text{Theo}}$$

$$J_W = A(\pi_{D,m} - \pi_{F,m})$$



$$J_W = A \left[\pi_{D,b} \exp(-J_W K) - \pi_{F,b} \exp\left(\frac{J_W}{k_F}\right) \right]$$

Characterizing the Support Layer

$$K = \frac{t \tau}{D \varepsilon}$$

K solute resistance to diffusion

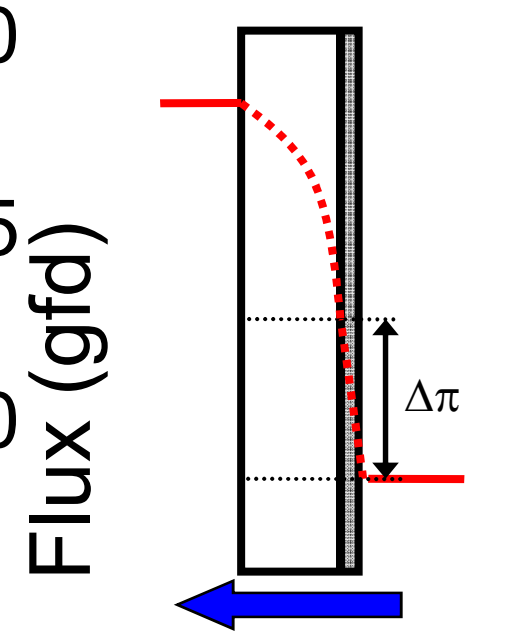
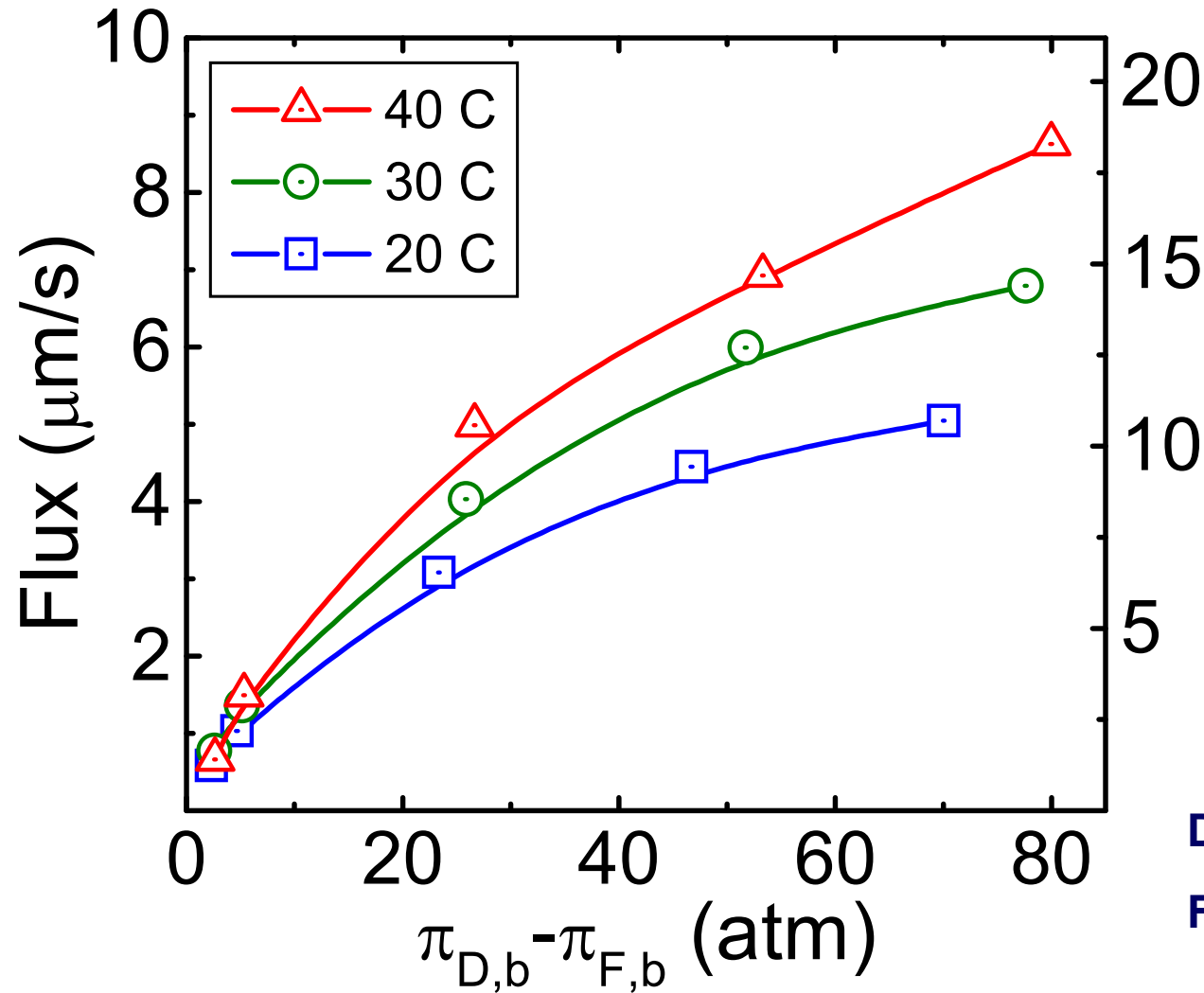
t support layer thickness

τ tortuosity

ε porosity

D draw solute diffusivity

Quantifying Internal CP

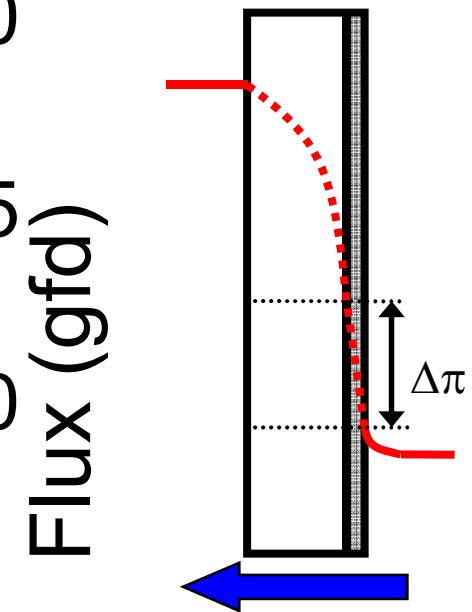
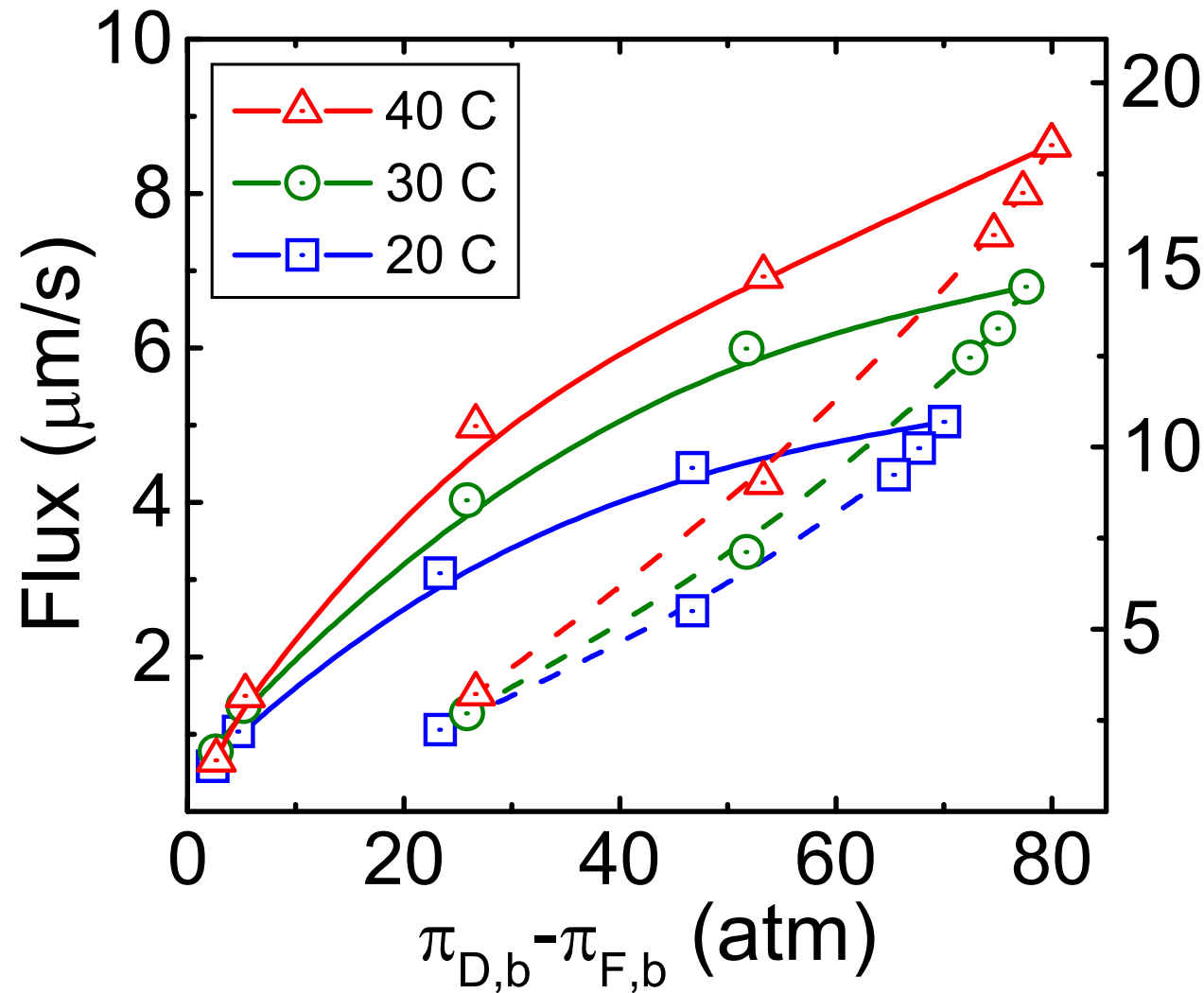


System Conditions

Draw: 0 to 1.5 M NaCl

Feed: Deionized water

Quantifying Internal CP



System Conditions

Draw: 1.5 M NaCl

Feed: 0 \rightarrow 1 M NaCl

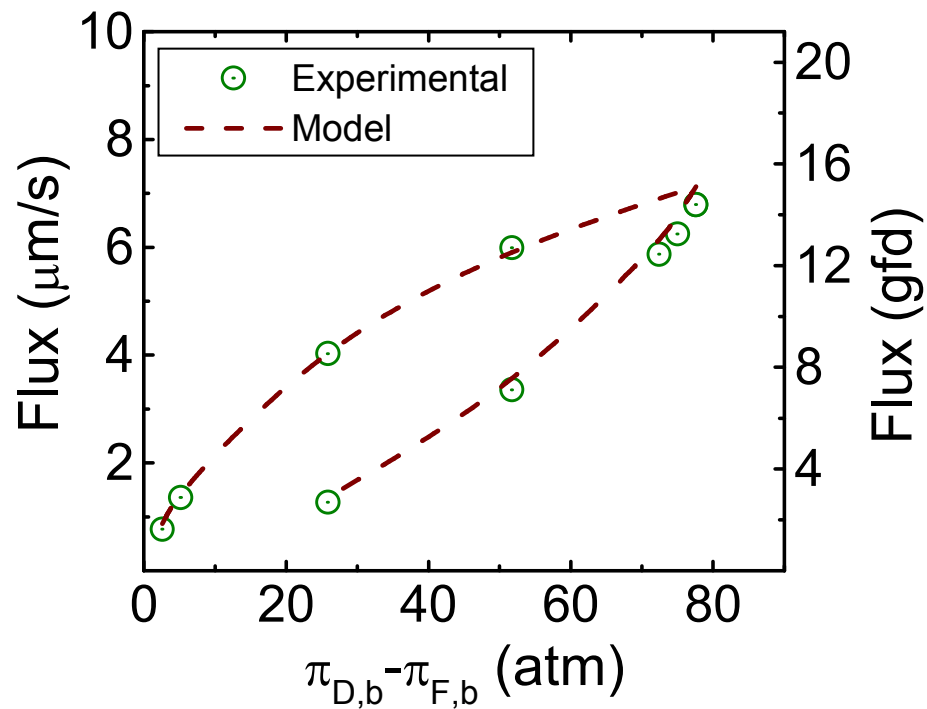
Modeling Internal CP

$$J_W = A \left[\pi_{D,b} \exp(-J_W K) - \pi_{F,b} \exp\left(\frac{J_W}{k_F}\right) \right]$$

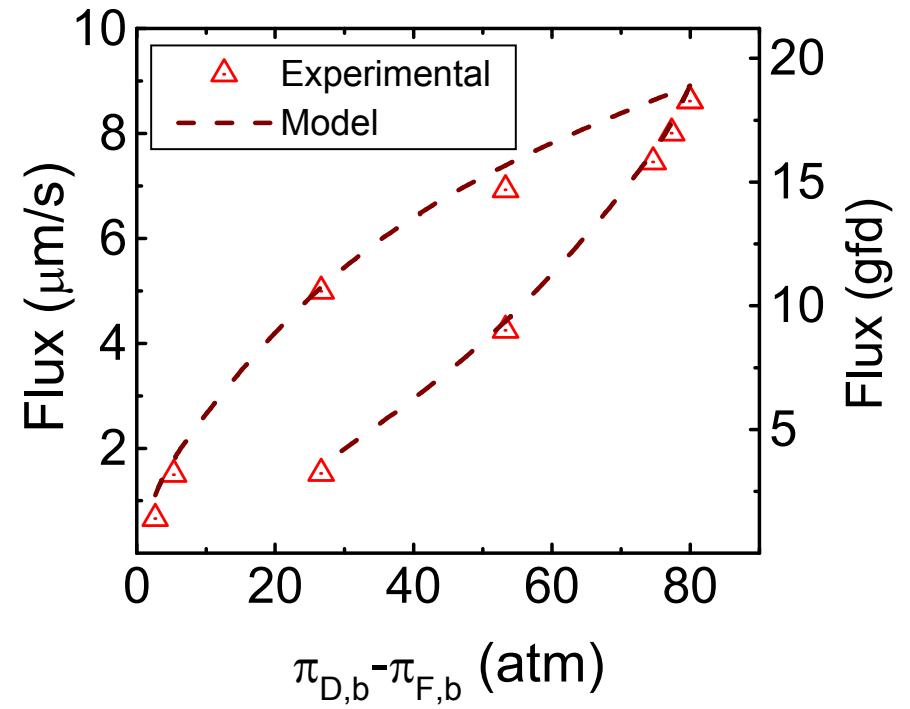
$$K = \frac{t\tau}{D\varepsilon}$$

Modeling Internal CP

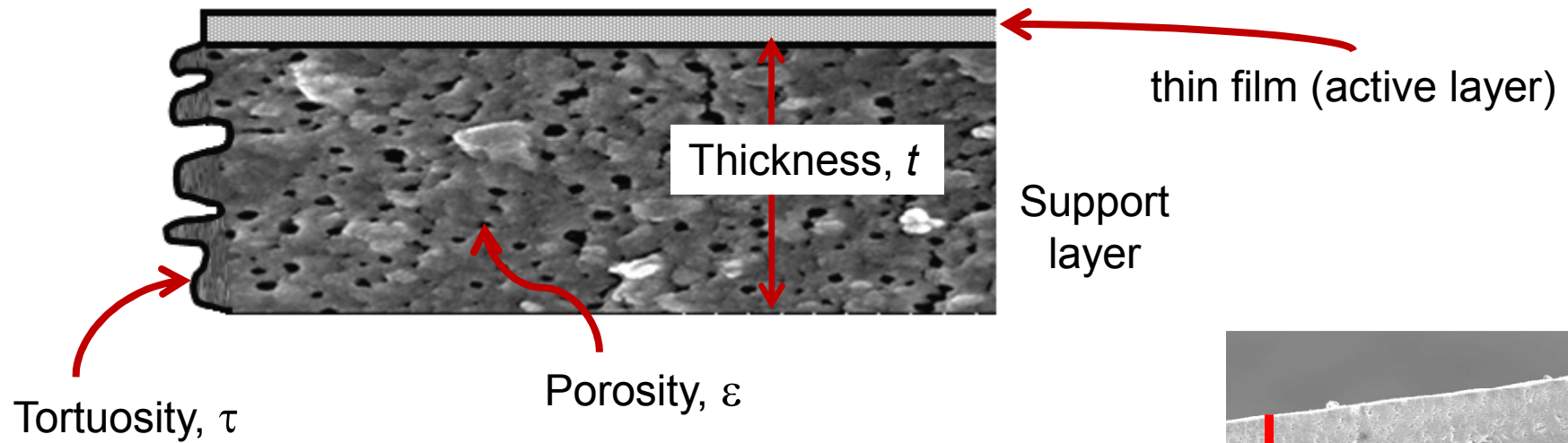
30 °C



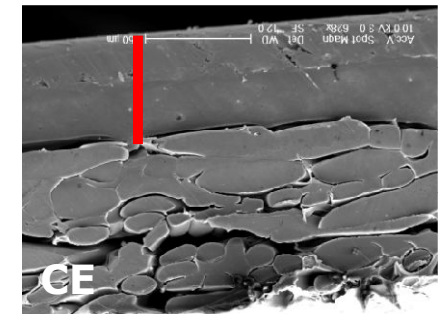
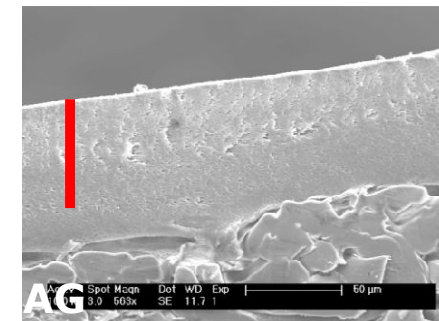
40 °C



Characterizing the Support Layer

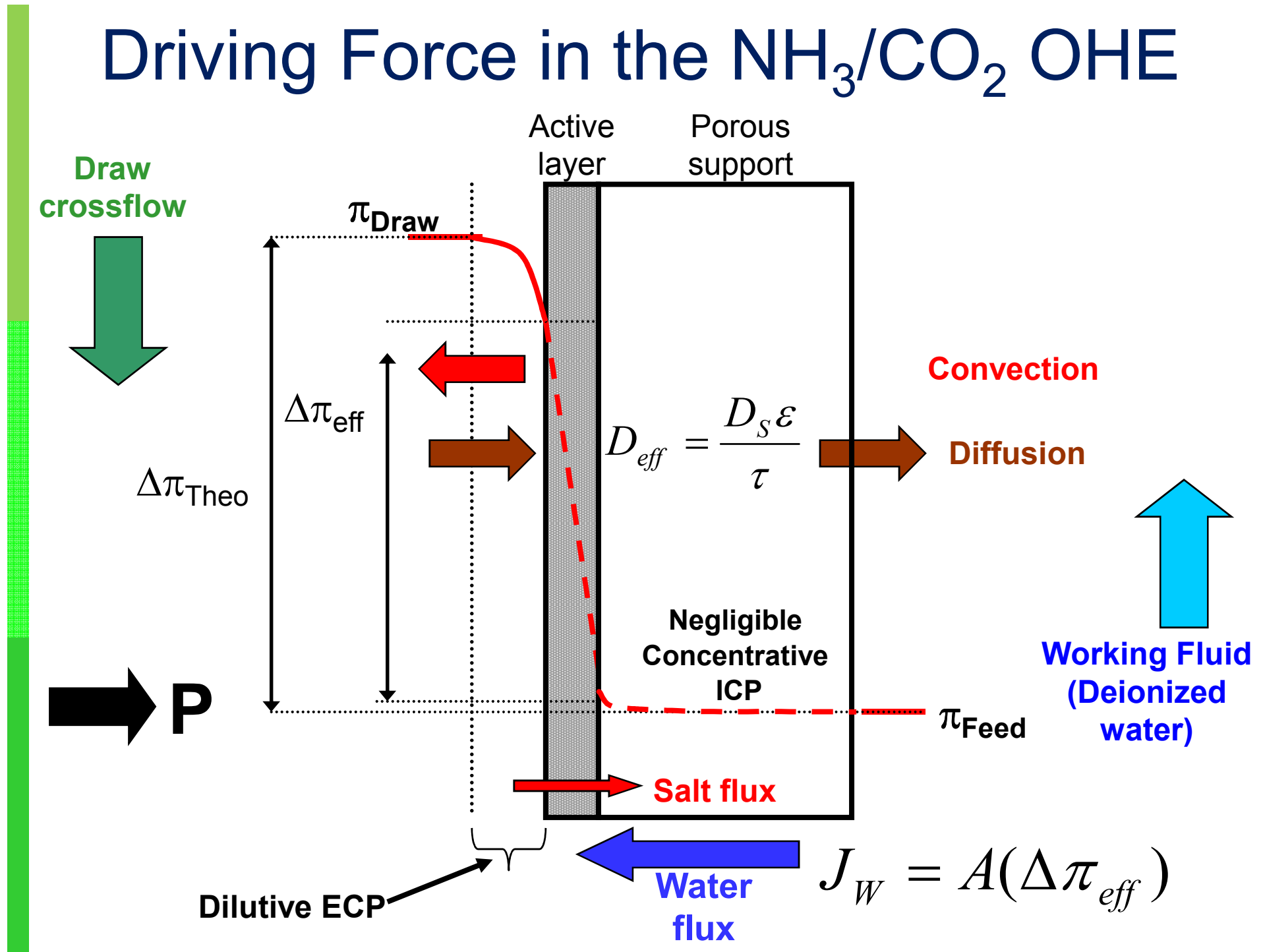


$$K = \frac{t \tau}{D \epsilon}$$

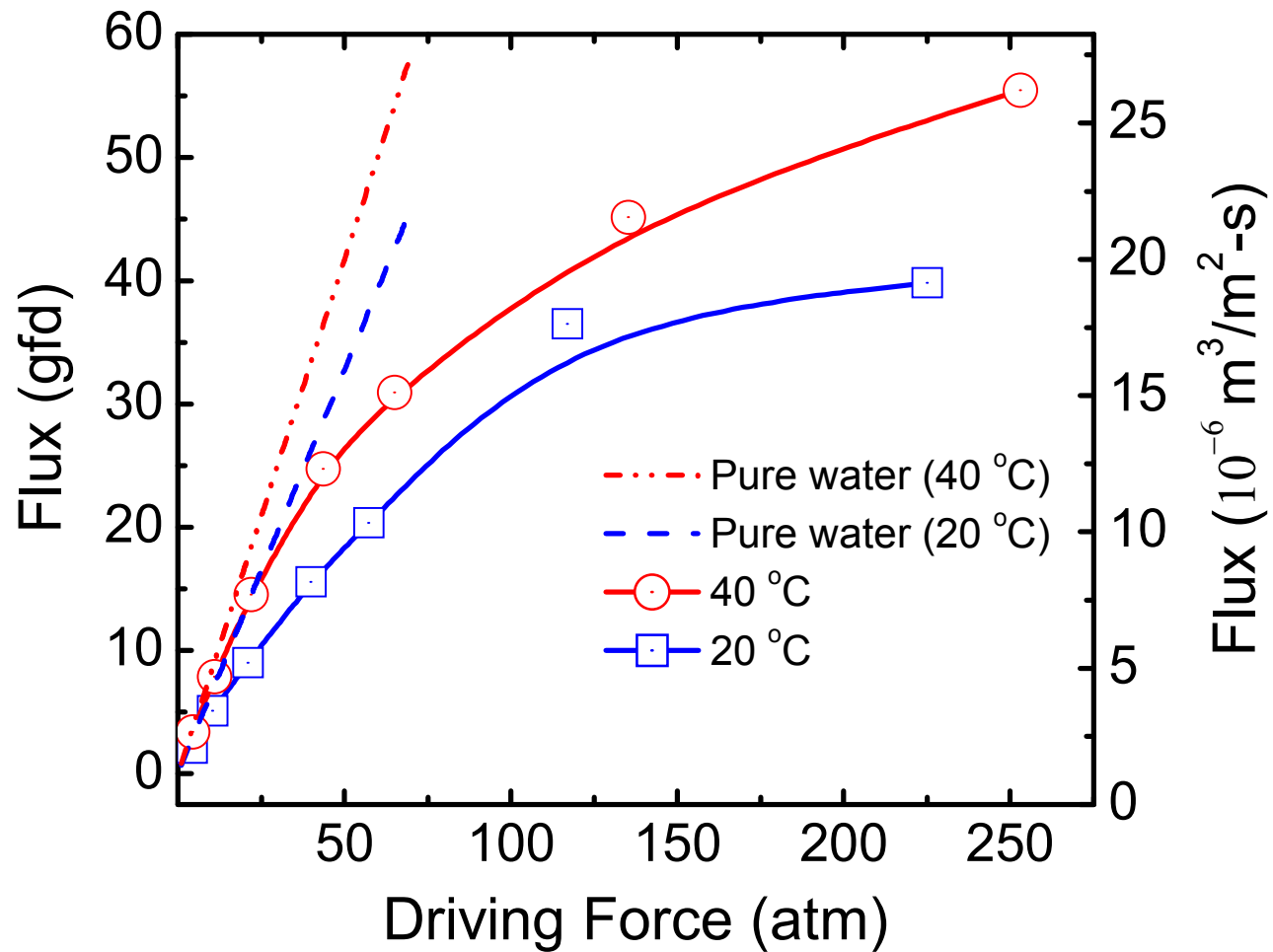


Red lines indicate 50 μm

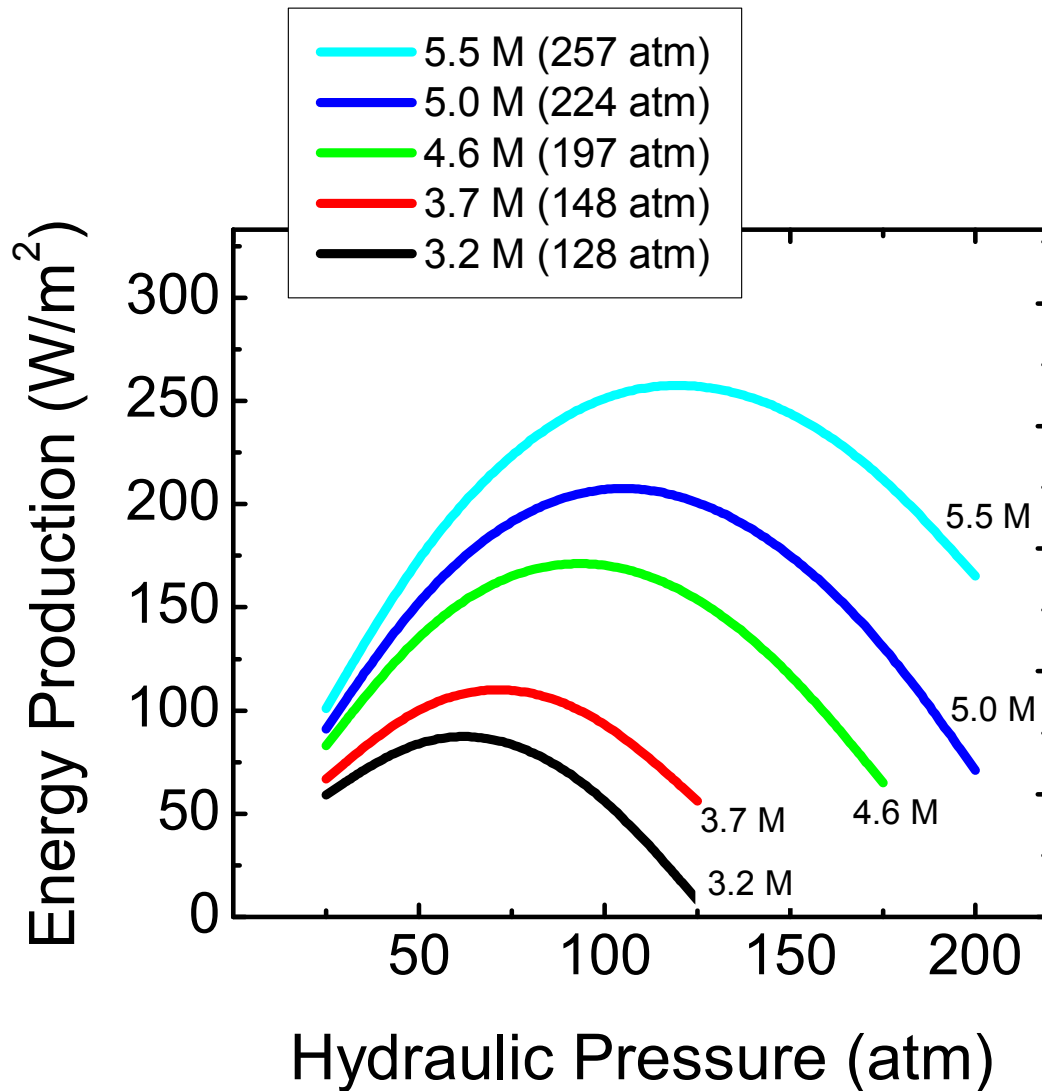
Driving Force in the NH₃/CO₂ OHE



Very High Water Fluxes



Very High Membrane Power Density

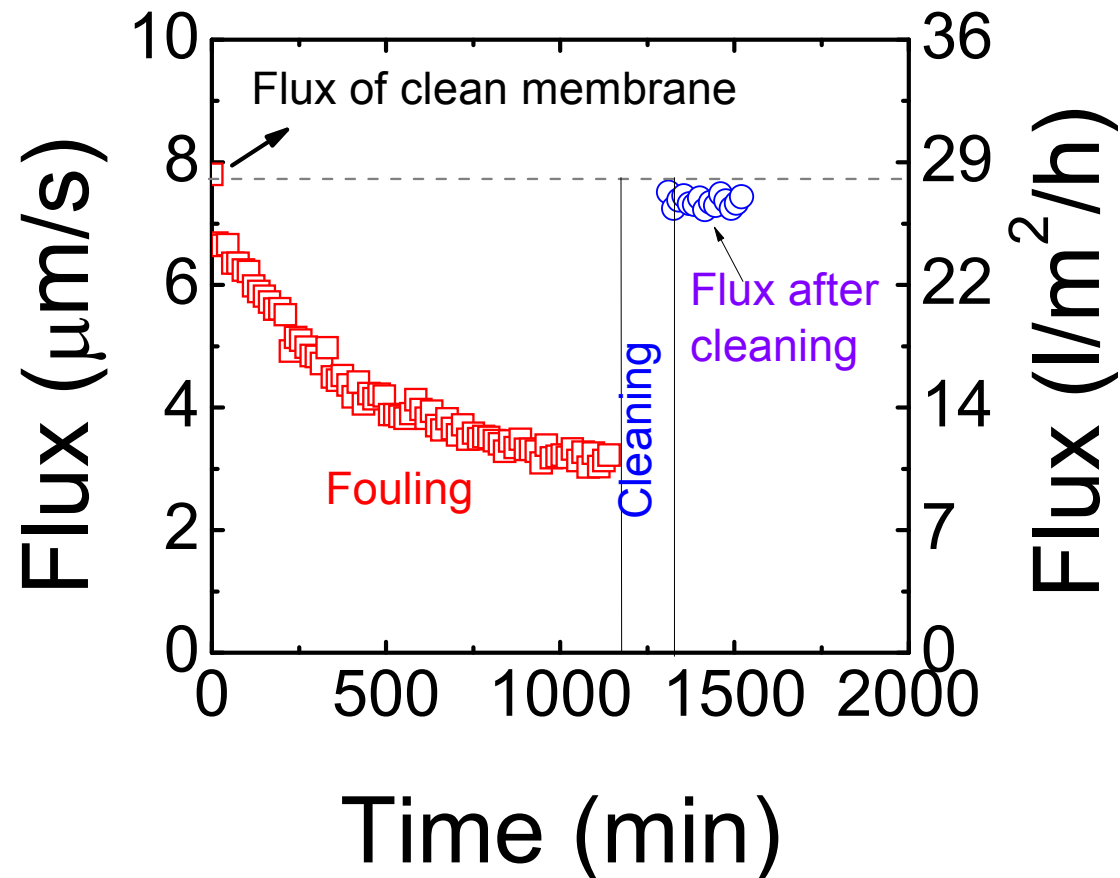


$$W = E\dot{V}\Delta P$$



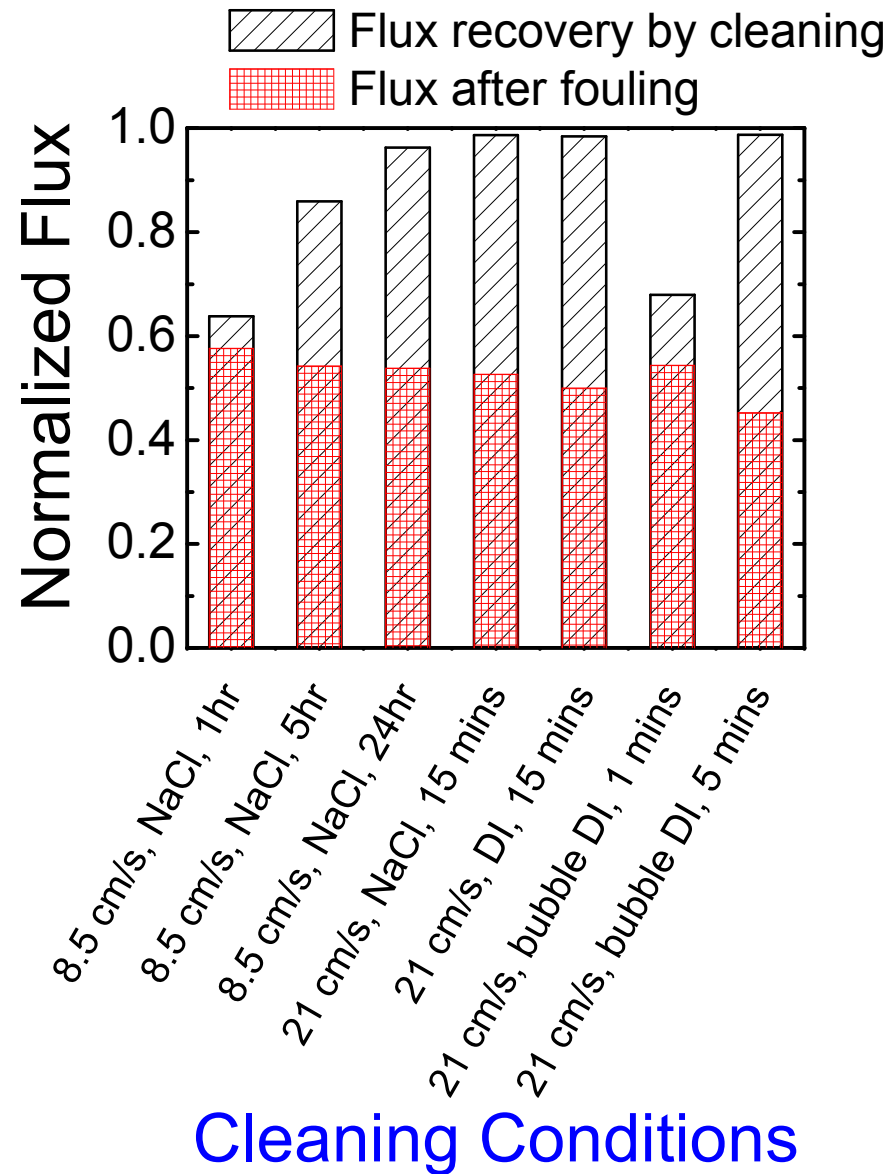
Fouling and Fouling Reversibility

Organic Fouling Reversibility in Forward Osmosis

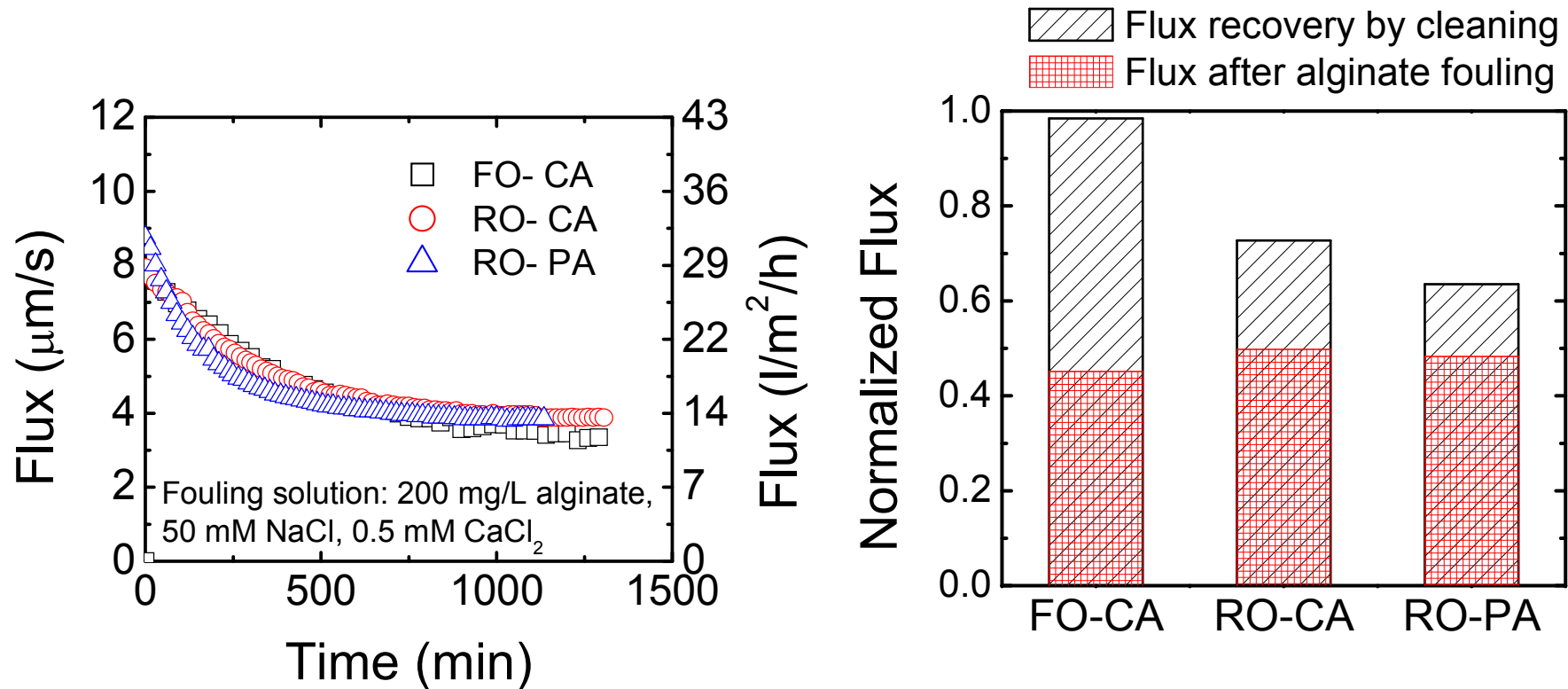


- FO membrane: CA (Hydration Tech)
- Organic foulant (200 mg/L alginate); 50 mM NaCl; 0.5 mM Ca²⁺
- Cleaning: 50 mM NaCl, 15 min

Organic Fouling Reversibility

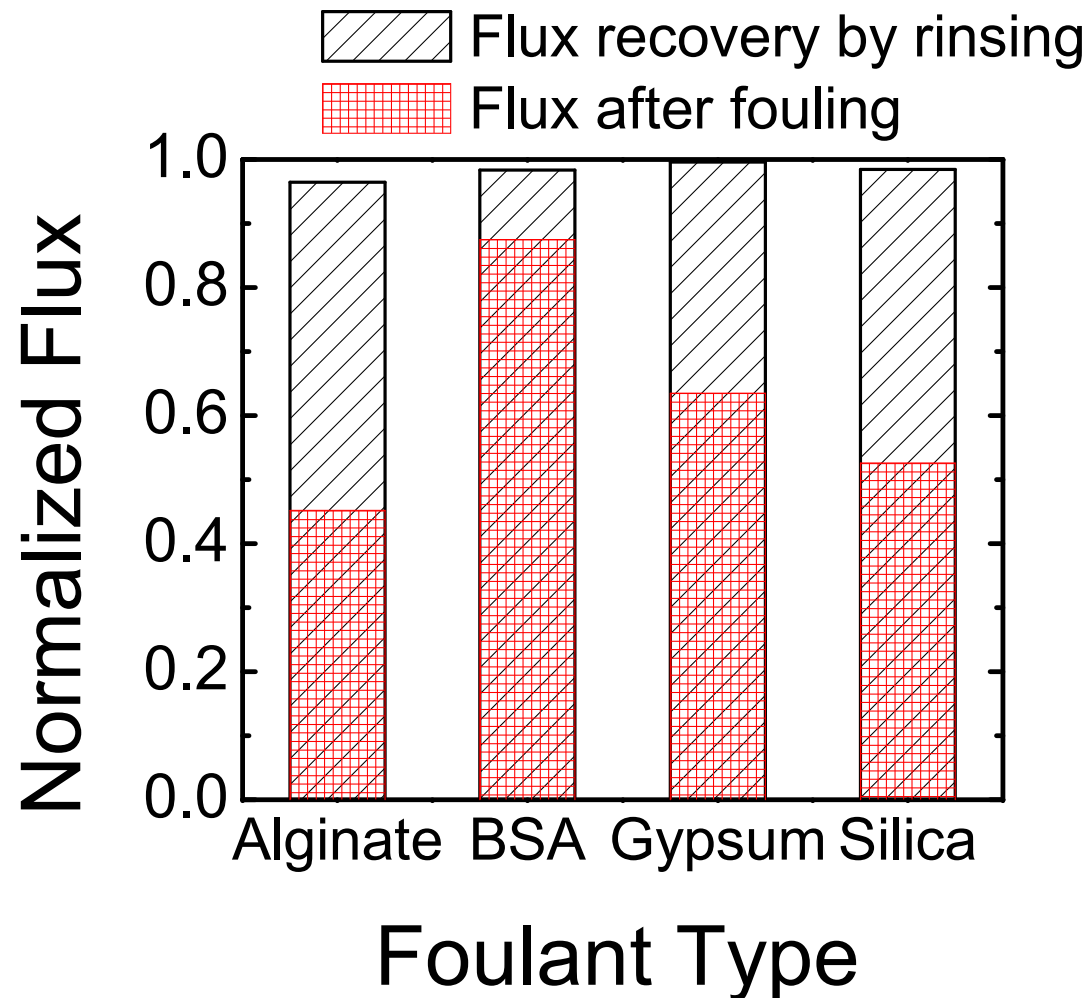


Organic Fouling Reversibility: FO versus RO



- FO – CA: CA membrane in FO mode
- RO – CA: CA membrane in RO mode (hydraulic pressure)
- RO – PA: Polyamide TFC membrane in RO mode

FO Exhibits Fouling Reversibility with a Wide Range of Foulants



Concluding Remarks

- Forward osmosis can be used as a standalone process or as part of an hybrid system (e.g. FO-RO)
- Forward osmosis is less prone to fouling; may use less prime (electric) energy
- Need to develop an appropriate membrane with low internal concentration polarization

Acknowledgments

- Graduate Students/Postdocs: Jeffrey McCutcheon, Robert McGinnis, Baoxia Mi, Sangyoup Lee
- World Class University (WCU) project (No. R33-2008-000-10046-0)
- The Ministry Education, Science, and Technology, Korea
- Korea Science and Education Foundation (KOSEF)
- US Office of Naval Research; NSF through WaterCAMPWS

